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spectroscopy is the study of the interaction of light and matter many types of spectroscopy rely on the ability of atoms and molecules to absorb or emit electromagnetic em radiation the absorption or emission of different forms of em radiation is related to different types of transitions spectroscopy is the study of how light interacts with matter we can use spectroscopy to determine the structure and functional groups in organic compounds we will be learning about how to use ir uv vis and nmr spectroscopy spectroscopy means the dispersion of light into component colours in simple words it is a method to measure how much light is absorbed by a chemical substance and at what intensity of light passes through it as per analytical science every element or compound has a unique characteristic spectrum oct 14 2021 spectroscopy is a complex art but it can be very useful in helping scientists understand how an object like a black hole neutron star or active galaxy is producing light how fast it is moving and even what elements it is made of a spectrum is simply a chart or a graph that shows the intensity of light being emitted over a range of energies spectroscopy pertains to the dispersion of an object s light into its component colors i e energies by performing this dissection and analysis of an object s light astronomers can infer the physical properties of that object such as temperature mass luminosity and composition spectroscopy is a branch of science concerned with the spectra of electromagnetic radiation as a function of its wavelength or frequency measured by spectrographic equipment and other techniques in order to obtain information concerning the structure and properties of matter 8 spectral measurement devices are referred to as spectrometers spectrophotometers jan 26 2023 spectroscopy study of the absorption and emission of light and other radiation by matter as related to

the dependence of these processes on the wavelength of the radiation more recently the definition has been expanded to include the study of the interactions between particles such as electrons protons and ions as well as their interaction with other particles spectroscopic techniques are widely used in forensic laboratories for quantitative and qualitative analysis this article provides an overview of the spectroscopic techniques most commonly encountered in forensic laboratories infrared spectroscopy raman spectroscopy x ray fluorescence scanning electron microscopy energy dispersive x ray spectroscopy and spectroscopy is used as a tool for studying the structures of atoms and molecules the large number of wavelengths emitted by these systems makes it possible to investigate their structures in detail including the electron configurations of ground and various excited states spectroscopy is the technique of splitting light or more precisely electromagnetic radiation into its constituent wavelengths a spectrum in much the same way as a prism splits light into a rainbow of colours however in general a spectrum is generally more than a simple rainbow of colours the energy levels of electrons in atoms and molecules are quantised and the

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